



NATIONAL OPEN UNIVERSITY OF NIGERIA
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ABUJA.

FACULTY OF SCIENCES
DEPARTMENT OF PURE AND APPLIED SCIENCE
SECOND SEMESTER EXAMINATION 2020..

COURSE CODE: CHM 391
COURSE TITLE: PRACTICAL CHEMISTRY V – INORGANIC AND ANALYTICAL
CREDIT UNIT: TWO (2)
TIME: 2 HOURS
INSTRUCTION: Answer question one and any other three questions.

QUESTION ONE

1a. Calculate the total alkalinity and total CO_2 of a water sample which was titrated with standard solution of 0.02M HCl and 0.02 Na_2CO_3 solution. The volume of Na_2CO_3 used for the total alkalinity and total CO_2 is 7.20 and 7.50 respectively 9 mks

1bi. Explain the term reprecipitation in precipitation gravimetry? 2 mks

1bii. Plot a graph of absorbance against concentration of the following:

	Concentration Mg/l	Absorbance
Calibration Zero	0.0000	0.0065
Standard 1	0.5000	0.0350
Standard 2	1.0000	0.0650
Standard 3	2.0000	0.1253

From the graph determine the concentration at an absorbance of 0.06. 7 mks

1c. List the different types of potentiometric titration you know. 4 mks

1d. Washing can be used to remove impurities, explain. 3 mks

QUESTION TWO

2a. What is homogeneous precipitation 4 mks

2b. Describe an experimental procedure for determining the concentration of cyclohexane using IR spectroscopy 5 mks

2c. Explain the methods used in acidity measurement 6 mks

QUESTION THREE

3. In qualitative analysis of cations, to separate and identify individual cations present in an unknown mixture containing a mixture of two or more cations, the unknown solutions or mixture was diluted using deionized water and addition of dilute HCl which gives a white precipitate. Now;

- i. Identify and separate the cations present 9 mks
- ii. State the confirmatory test of the cations present. 6 mks

QUESTION FOUR

- 4a. There is an assertion that not all molecules can absorb in the infrared region. Discuss $4\frac{1}{2}$ mks
- 4b. Explain the simple ball-stick diagram using AgCl $4\frac{1}{2}$ mks
- 4c. State the economic importance of hard water 3 mks
- 4d. Mention the criteria for a complex to exhibit charge-transfer behaviour. 3mks

QUESTION FIVE

- 5a. Calculate the total hardness of a water sample if 25.0ml of 0.01M EDTA was used to titrate 10ml of a water sample. $5\frac{1}{2}$ mks
- 5b. Mention the main issues involved in analyzing a sample using infrared spectroscopy $4\frac{1}{2}$ mks
- 5c. State the energy activities between HOMO and LUMO in UV-Visible radiation 5 mks