



NATIONAL OPEN UNIVERSITY OF NIGERIA
PLOT 91, CADASTRAL ZONE, NNAMDI AZIKIWE EXPRESSWAY, JABI - ABUJA
FACULTY OF SCIENCES
DEPARTMENT OF PURE & APPLIED SCIENCES
OCTOBER/NOVEMBER 2019_2 EXAMINATION

COURSE CODE: CHM 424

COURSE TITLE: NON AQUEOUS SOLVENTS

CREDIT: 2 UNIT

TIME ALLOWED: 2 HOURS

Instruction: Answer **question 1** and any other three.

- Q1a What is induced dipole induced interaction **(5 marks)**
- b. Briefly discuss the effect of dielectric constant and solubility on Arsenic chloride. **(4 marks)**
- c. Outline any two advantages of liquid ammonia as solvent over water **(4 marks)**
- d. Give properties of phosphorylchloride **(2 marks)**
- e. Highlight the general characteristics of polarprotic solvent. Give two examples? **(4 marks)**
- f. Write the equation for the following reaction of POCl_3 , autoionization, reaction with triethylmine and with IronIIIchloride. **(6 marks)**
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- 2a. Enumerate the characters of dinitrogentetroxide that can be used as medium for conducting chemical reactions **(5 marks)**
- b. Give the physical properties of liquid N_2O_4 **(5 marks)**
- c. With typical equations represent the reactions of N_2O_4 with lithium, sodium, aluminum and zinc nitrate **(5 marks)**
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- 3a. With appropriate equations represent the solvolytic reactions of N_2O_4 with $(\text{C}_2\text{H}_5)_2\text{NH}_2\text{Cl}$, MCl , $(\text{MgCH}_2\text{O})_6\text{Cl}_2$, $(\text{Mg}(\text{ClO}_4)_2)$ and Li_2CO_3 **(5 marks)**
- b. Using suitable equations represent the adducts formation of N_2O_4 with inorganic compounds and comment on the stability of the product.

- (5 marks)**
- c. Enumerate the special features of liquid SO_2 as solvent **(5 marks)**
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- 4a. Draw the resonance structure of SO_2 **(2 marks)**
- b. How does SO_2 undergo autoionization? **(6 marks)**
- c. With appropriate equations represent the neutralization reactions of SO_2 **(7 marks).**
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- 5a. What is metathetical reactions? **(6 marks)**
- b. How is amphoteric reactions in liquid sulphate analogous to amphoteric substance in water ? **(6 marks)**
- c. With equations only, represent the complex reaction of antimony chloride **(3 marks).**